

Highly Enantioselective 6π Photoelectrocyclizations Engineered by Hydrogen Bonding

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Introduction

Article



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•Vordiplom (BS), Heinrich-Heine Universität Düsseldorf (1991-1995)
•PhD, UNC Chapel Hill (Cynthia K. Schauer) (1995-2000)
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Computational chemistry, focusing on the mechanism of organometallic catalysts
Seeks to predict new catalysts



Highly Enantioselective 6π Photoelectrocyclizations Engineered by Hydrogen Bonding

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ABSTRACT: Photochemical electrocyclization reactions are valued for both their ability to produce structurally complex molecules and their central role in elucidating fundamental mechanistic principles of photochemistry. We present herein a highly enantioselective 6π photoelectrocyclization catalyzed by a chiral Ir(III) photosensitizer. This transformation was successfully realized by engineering a strong hydrogen-bonding interaction between a pyrazole moiety on the catalyst and a basic imidazolyl ketone on the substrate. To shed light on the origin of stereoinduction, we conducted a comprehensive investigation combining experimental and computational mechanistic studies. Results from density functional theory calculations underscore the crucial role played by the prochirality and the torquoselectivity in



the electrocyclization process as well as the steric demand in the subsequent [1,4]-H shift step. Our findings not only offer valuable guidance for developing chiral photocatalysts but also serve as a significant reference for achieving high levels of enantioselectivity in the 6 σ photoelectrocyclization reaction.



Tehshik Yoon

•AB, Harvard (David A. Evans) (1992-1996)
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•PhD, CalTech / UC Berkeley (David W.C. MacMillan) (2000-2002; 1998-2000)
•Postdoc, Harvard (Eric N. Jacobsen) (2002-2005)

•Faculty at UW Madison since 2005

•Methodology-driven, specializing in the controlled synthesis of complex molecules via photochemistry

Context

Thermal: 6π disrotatory Magomedov & co-workers (2004) *t-*Bu *t-*Bu н Ph MAD (0.5 eq.) Ts C Ph Ph NMP (0.25 eq.) Me Ts Ts PhMe, RT t-Bu t-B OMAD MAD 91% single diastereomer Bergman, Trauner, Hsung & co-workers (2008) t-Bu *t-*Bu & co-workers (2010) n -Sc-Ph (OTf)₃ ,Η (1 eq.) (0.7 eq.) Ph Ρĥ R* (CDCl₂)₂, 5 h, RT **R**²* **R**²* Cp₂ZrMe₂, (0.5 eq.) O, X = O, NH, CH₂; *n* = 0, 1 57%-77% ee up to 96:8 d.r. chiral R* or R^{2*} Photochemical Chapman & co-workers (1968) Gong, Meggers, & co-workers (2015) O R² t-Bu chiral cat. hν ŃR (2-5 mol%) Et₂O, RT Ĥ JR Me Ňе \mathbb{R}^2 R R³ 77% yield ex: R_1 , R_2 , Ar = Ph, $R_3 = Me$ A-RhO: 80% yield, 96% ee 'Ar

Previous Photocatalytic Electrocyclizations





Paton, Smith, & co-workers (2017)





Edtmüller, V.; Pöthig, A.; Bach, T. *Tetrahedron* **2017**, *73*, 5038 Münster, N.; Parker, N. A.; van Dijk, L.; Paton, R. S.; Smith, M. D. Angew. Chem. Int. Ed. **2017**, *56*, 9468 Jones, B. A.; Solon, P.; Popescu, M. V.; Du, J.-Y.; Paton, R. S.; Smith, M. D. J. Am. Chem. Soc. **2023**, *145*, 171

Reaction Optimization

entry	modification	yield	d.r.	ee
1	PhMe	57%	6:1	87%
2	CH ₂ Cl ₂	29%	3:1	87%
3	MeCN	17%	3:1	53%
4	20 °C	60%	8:1	73%
5	-40 °C	58%	5:1	96%
6	1 mol% Λ-PC	39%	5:1	74%
7	0.01 M	55%	7:1	82%
8	in air	<5%		-
9	dark	0%	-	_
10	no Λ-PC	0%		_
11	(Z)- 9	84%	8:1	91%
13	11 instead of 9	0%	-	-
14	12 instead of 9	0%	-	-
15	13 instead of 9	0%	-	-





 A-PC serves both as a photosensitizer and as a chiral ligand to enhance enantioselectivity

- •Presence of a **carbamate** (with Boc or Fmoc) is required for the reaction to work
- •The PC can only transfer the energy to the substrate with a carbamate
- •Cyclized product from tertiary amine **12** decomposes
- •Triplet energies based on calculations: •carbamate: 50.9 kcal·mol⁻¹ •secondary amine: 53.4 kcal·mol⁻¹
- •ether: 54.0 kcal·mol⁻¹
- -Presence of the imidazole (NMI) allows for hydrogen bonding between $\Lambda\text{-PC}$ and the substrate
- •Binding of the substrate to the photocatalyst post energy transfer promotes the enantioselectivity

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Substrate Scope



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Incompatible Substrates



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Mechanistic Inquiry



A) Reaction profile monitoring the concentrations of (*E*) & (*Z*) olefin starting materials, as well as *cis* & *trans* products

B) Stern-Volmer plot, showing quenching of Λ-PC via both (*E*) & (*Z*) olefins
likely through energy transfer (oxidation & reduction potentials of Λ-PC are too low to perform either redox process)





substrate and Λ -PC results in lower ee

Performing [2+2] cycloaddition



•The presence of NMI in enone **28** performs with relatively good ee (up to 91% ee at -65 °C) in the [2+2] cycloaddition

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Proposed Mechanism



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Stereodetermining Factors



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Computational Studies



Concluding Remarks



•Effective method to access indoline scaffolds with high ee, and reasonable to high yield on a variety of substrates

•Photocatalyst serves two important purposes:

- •Chiral ligand (effectuated through hydrogen bonding)
- •Photosensitizer which performs an energy transfer to the substrate

•Considerations from this method:

•NMI (or potentially another strong H-bond donor) is required to achieve high ee

•Carbamate is required for the reaction to work (incompatible with secondary amine and ether, and tertiary amine decomposes)

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